Safety Data Sheet Palladium Sulfate Solution

Palladium Sulfate (II) Solution

Palladium plating and catalysts

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As above

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1. Chemical Articles and Company Information

Name of Chemical Article: Company Name: Address: Tel.: Fax: Emergency Contact: Recommended Applications and Use Restrictions:

2. Summary of Hazards

GHS Classification		
Damage to health	Acute (oral) toxicity	Class 5
	Acute (inhaled: dust and mist) toxicity	Class 2
	Skin corrosiveness and irritation	Class 1A
	Critical injury to eyes and eye irritant	Class 1
	Specific marker organ and systemic	
	toxicity (single exposure)	Class 1 (respiratory system)
	Specific marker organ and systemic	
	toxicity (repeated exposure)	Class 1 (respiratory system)
Damage to the environment	Acute aquatic environmental harm	Class 3
GHS Label Elements		\sim
Picture descriptions:		
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Cautionary terms:	Danger	
Hazard information:	Harmful if ingested (orally)	
	Dangerous to life if inhaled (mist)	
	Critical chemical burns to skin and injury to	eyes
	Risk of malign influence on reproductive fu	nctions or fetus
	Risk of organ (haemal system) damage and	respiratory organ irritation
	Risk of organ (dental and respiratory system	n) damage due to long-term or repeated exposure
	Risk of toxicity to aquatic life	
	Risk of toxicity to aquatic life forms due to	long-term effects
Cautions		
Safety Measures:	Do not handle until all safety precautions an	d readings are understood.
	When using the product, do not eat, drink, o	r smoke.
	Store away from reducing agents.	
	Avoid heat.	
	Wear protective gloves, goggles, and face m	nask.
	Prevent contact with eyes, skin, and clothing	g.
	Do not inhale the dust.	
	Wash hands thoroughly after handling.	
	Avoid discharging into the environment.	

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Emergency Measures:	If the substance contacts the eye, irrigate with water thoroughly for several minutes. If contact lenses can be removed easily, remove and wash them.
	If eye irritation persists, consult a physician and receive treatment.
	If clothing is spattered, promptly remove and isolate all soiled clothing.
	If the substance adheres to the skin, wash using copious amounts of soap and water.
	If skin irritation occurs, consult a physician to receive diagnosis and treatment.
	If exposed or fear exposure, consult a physician to receive diagnosis and treatment.
	If ingested, rinse out the mouth, and immediately consult a physician to receive diagnosis and treatment.
	If you feel unwell, consult a physician to receive diagnosis and treatment.
	Gather any leaks.
Storage:	Lock the storage location.
	Store away from reducing agents.
Disposal:	If discarding contents or containers, entrust to a specialized waste disposal company.
Other hazards:	Not available

3. Composition and Component Information

Single Substance Chemical name	e or Mixture: or general name:	Single Substance Palladium Sulfate (II)
Another name:		-
Concentration of	concentration range:	PdSO ₄ : 7.6% (Pd 4.0%)
Molecular form	ıla (molecular weight):	PdSO ₄ (202.48)
Chemical charac (rational or struc		
CAS No.:		13566-03-5
Reference numb in japan(CSCL a	ers in gazetted list ind ISHL):	1-(3)-375
Impurities and s additives that co classification:		Sulfuric acid
4. Emergency M	easures	
If inhaled:		Move to a location with fresh air. If you feel unwell, consult a physician to receive diagnosis and treatment.
Adhesion to skir	.:	Remove soiled clothing and shoes, and thoroughly wash any adhesions or contact parts with cold or lukewarm water. If changes in appearance manifest, or pain continues, consult a physician.
Contact with eye	25:	If the substance contacts the eye, irrigate with water thoroughly for several minutes. If eye irritation persists, consult a physician and receive treatment.
If ingested:		Rinse mouth.
-	ant sign of an acute e tardive symptom and	If you feel unwell, consult a physician to receive diagnosis and treatment. Corrosiveness, burning sensation, pharyngeal pain, cough, feeling of dyspnea, shortness of breath, rubefaction, pain, blisters, severe skin burns, stomach pains, shock, or lethargy.
Protection of peo emergency meas	ople implementing ures:	Rescuers should wear suitable protective equipment according to the circumstances.
	ons for physicians:	In many cases, pulmonary edema symptoms are not identified until 2 to 3 hours have elapsed, so rest and continual observation are necessary.

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5. Measures during Fires	
Extinguishing agents:	This product in itself does not burn. Use the appropriate extinguishant depending on neighboring fires.
Extinguishing that must not be used:	No information
Characteristic dangers:	Non-flammable, so itself is not burned, but strong heat causes harmful gas (SOX) to be emitted, so wear protective equipment when firefighting.
Characteristic extinguishing methods:	Promptly move containers in the vicinity of the fire to a safe location. If moving is not possible, scatter water on the containers and their surroundings to cool. If ignition occurs, douse the fire using copious amounts of water. At this time, make sure there is no discharge of concentrated effluent into rivers, etc.
Protection of firefighters:	Fight fire from upwind, and avoid inhaling poisonous gasses, etc.
	Wear respiratory protectors depending on the circumstances.
	Keep upwind all personnel unnecessary to disaster prevention activities.
6. Measures during Leaks	
Physical precautions, protective equipment, and measures during emergencies:	The worker wears a tool for appropriate protection (in item of "8. Exposure Avoidance and Protection Measures" reference) and avoids clothes, contact and inhalation to skin. Cordon off the periphery of the dispersal area to prohibit the entrance of personnel.
Environmental precautions:	Avoid discharging into the environment.
Methods and materials for contamination and methods and	
materials for cleaning up:	No information
Collection and neutralization:	Collect spatter in a container, and neutralize using soda ash, and then wash away using copious amounts of water.
Preventing secondary accidents:	Prevent inflow to drainage ditches, sewers, cellars, or sealed locations.

7. Handling and Storage Precautions

Handling	
Technical measures:	Install local exhausters, and eye and hand washing facilities, in the handling locations.
	Ideally, handle in locations with local exhausters and overall ventilators.
	Wear protective gloves, goggles, and face mask.
Precautions for safe handling:	Do not handle until all safety precautions and readings are understood.
	When using the product, do not eat, drink, or smoke.
	Do not touch, inhale, or drink.
	Use only outdoors or in a well-ventilated area.
	Wash hands thoroughly after handling.
	If not using deliberately, avoid discharging into the environment.
Contact evasion:	In item of "10. Stability and Reactivity" reference.
Storage	
Safe storage conditions:	Avoid contact with heat, strong alkalis, and reducing agents.
	Lock the storage location.
	Securely seal the containers, and store in a well-ventilated.
	Technical measures:No information
Container and packing materials:	Use containers regulated by UN transportation laws.

8. Exposure Avoidance and Protection Measures

Control concentration:	No information
Tolerable concentration:	
Japan Society for Occupational Health (2015)	1 mg/m ³ (As the sulfuric acid)
ACGIH (2013)	0.2 mg/m^3 (TLV-TWA) (As the sulfuric acid)
Equipment Measures:	Install local exhausters, and eye and hand washing facilities, in the handling locations.

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Protective Equipment	
Respirator:	Wear suitable respirator. (Breathing apparatuses,etc)
Hand protective equipment:	Wear suitable protective gloves. (Neoprene gloves, etc.)
Eye protective equipment:	Wear suitable eye protective equipment. (Goggles, etc.)
Skin and body protective equipment:	Wear suitable protective face equipment, clothing, and protective shoes, etc.
	(Impermeable protective clothing, protective boots, etc.)

9. Physical and Chemical Properties

7. I hysical and Chemieal I topertie.	5
Physical properties	
Shape:	Liquid
Color:	Reddish-brown
Odor:	Odorless
Odor threshold value:	No information
pH:	1 max.
Melting point and coagulation point:	No information
Boiling point, initial boiling point, and boiling range:	No information
Ignition Point:	No information
Vaporization speed (butyl chloride=1):	No information
Burnability (solids and gas):	No information
Explosion range:	No information
Vapor pressure:	No information
Vapor density (vapor=1):	No information
Specific gravity (density):	1.2~1.3
Solubility:	No information
n-Octanol/water partition coefficient:	No information
Spontaneous ignition temperature:	No information
Dissolution temperature:	No information
Viscosity:	No information
10. Stability and Reactivity	
Reactivity:	No information
Stability:	When water is enough, it is disintegrated, and palladium(II)Oxide occurs.
Possibility of harmful reactions:	Reduced to metal palladium by a strongly reducing agent. Contact with metals and exposure to high temperatures causes a harmful gas (SOx) to be
	emitted.
	Contact or mixture with flammable substances may cause heating and ignition due to the catalytic reaction of palladium.
Conditions to be avoided:	Heat and Humidity
Incompatible substances:	Reducing agents, metals, strong alkalis, and organic substances
Hazardous degradable organisms:	Sulfur oxides
11. Harmfulness Information	
Acute toxicity	
Oral:	Rat oral: LD50; 1,420mg/kg. (As the sulfuric acid)
Pass; skin:	No data available
Inhalation:Gas	The definition of GHS is a liquid.
Inhalation:Steam	The definition of GHS is a liquid.
Inhalation:Dust,Mist	Inhalation (mist): Rat LC50; 0.375mg/L. (As the sulfuric acid)
Skin corrosiveness and irritation:	As the pH is 1 max., the substance is judged to be corrosive.
Critical injury to eyes and eye irritant:	The serious damage of eyes with the dissolution of the anterior chamber of eye was
	accepted in the accident example in the Homo sapiens. (As the sulfuric acid) Acridity was recognized for the eyes of the rabbit. (As the sulfuric acid)
	Action y was recognized for the eyes of the fabbil. (As the sufface actu)

Respiratory organ sensitivity:	
Skin sensitivity:	

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No data available No data available

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Germ-cell mutagenicity:	No data available
Carcinogenicity:	No data available
Reproductive toxicity:	No data available
Specific marker organs and systemic toxicity (single exposure):	Respiratory tract stimulation symptoms such as a cough, the shortness of breath are recognized in the inhalational revelation with the low concentration in the Homo sapiens. (As the sulfuric acid) The permanent influence such as the functional decline of the lungs and fiberization, the emphysema was recognized other than acute influence such as a cough, shortness of breath, the bloody phlegm discharge in the highly-concentrated revelation. (As the sulfuric acid) The bleeding of the lungs in the inhalational revelation and functional disorder were accepted in a marmot. (As the sulfuric acid)
Specific marker organs and systemic toxicity (repeated exposures):	Cell proliferation was accepted in the inhalational revelation examination for 28 days in rats by a larynx mucous membrane. (As the sulfuric acid) The nasal septum edema, pulmonary emphysema, atelectasis, bronchiolar hyperemia, edema, bleeding, respiratory tracts such as clots and lungs disorder was accepted in the repetition inhalational revelation examination for 14-139 days in the marmot. (As the sulfuric acid) The histologic changes such as the hyperplasia of the cell, the thickening of the wall were accepted in the inhalational revelation examination for 78 weeks in the Macaca fascicularis by the bronchiole of the lungs. (As the sulfuric acid)
Inhalable respiratory organ harmfulness:	No data available

12. Environmental Impact Information

Ecotoxicity	
Aquatic environmental harm (acute hazard):	Bluegills: LC50; 16 - 28 mg/L 96h (As the sulfuric acid)
Aquatic environmental harm	
(long-term hazard):	No data available
Hazard to the ozone layer:	The materials concerned are not listed by an affiliated book of Montreal Protocol.
13. Disposal Precautions:	
Residual waste:	Collect the palladium using a roasting and reduction process, or an oxide precipitation process. During roasting, gasses containing harmful substances are emitted, so do not roast using an
	incinerator without cleaning equipment. (Ideally, entrust to a specialized company.)
Dirty containers and packaging:	Suitably process containers according to the relevant regulations and local government standards.
	When disposing of empty containers, make sure to discard the contents completely.
14. Shipping Precautions International Regulations	
UN No.:	0707
	2796
Proper Shipping Name:	Sulphuric acid with 51% or less acid
Class:	Class 8: Corrosive
Sub Risk:	-
Packing Group:	II
Marine Pollutant:	-
Transport in bulk according to	
Annex II of MARPOL 73/78 and the IBC Code:	No
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Japanese Regulations	
Land Regulation Information:	Obey Poisonous and Deleterious Substances Control Law and Fire Services Act regulations.
Maritime Regulation Information:	Obey Ship Safety Law regulations.
Aviation Regulation Information:	Obey the Civil Aeronautics Act.

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Special safety measures:	During transport, avoid direct sunlight, and load so that the containers are not damaged, corroded, or leaking, and secure the load to prevent toppling.
	Do not transport together with food or livestock feed.
	Yellow card display is required during transport.
15. Applicable Laws	
Fire Service Act:	Substances requiring notification of storage (As the sulfuric acid) (200kg)
Poisonous and Deleterious	
Substances Control Law:	Non-medical deleterious substances
Industrial Safety and Health Law:	Notifiable substance (Article 57-2, government ordinance Article 18-2 attached table No. 9-613) (As the sulfuric acid)
Ordinance on Prevention of Hazards due	
to Specified Chemical Substances:	Specified Chemical Substance (Group 3) (As the sulfuric acid)
Regulations for the carriage and storage of	f
dangerous goods in ship:	Corrosive Substances
Civil Aeronautics Act:	Corrosive Substances
16. Other Information	
Bibliography:	GHS classification results database: NITE website
	GHS model SDS information: JISHA website
	Reagent guidebook (Revised 2003)
	Collection of Poisonous Materials Standard Notifications
	Dictionary of Chemistry (1987 30th printing: Kyoritsu Shuppan)
	16112 Chemical Products (2012 The Chemical Daily)
	Torra Chemical Froducts (2012 The Chemical Dury)

*Caution:

Hazard and harmfulness evaluations were created using the data and information available at the current time, but is not necessarily thorough, so handle with care.

Further, the data and evaluations described herein are not in any way guaranteed. The descriptions refer to normal handling, so for special handling, first implement safety measures conforming to the new application and methods of use.

This SDS is translated into English.(Original version is Japanese)